



Faculty Name: Dr. Rupa Pal

PROGRAM: Electronics & Communication Engineering	DEGREE: B. Tech
COURSE TITLE : Chemistry-I	SEMESTER: 1st
COURSE CODE : BS-CH101(C102)	SECTION: A
CATEGORY : Basic Science	CREDITS: 4
CONTACT HOURS/WEEK: 3(Lectures) +1 (Tutorial)	

MODEL LESSON PLAN

Unit	Name of the Topic	CO's	No. of Lecture
1	<i>Atomic and molecular Structure</i>	CO1	10
2	<i>Spectroscopic techniques and applications</i>	CO3	8
3	<i>Intermolecular forces</i>	CO1	4
4	<i>Use of free energy in chemical equilibrium</i>	CO2	8
5	<i>Periodic properties</i>	CO4	4
6	<i>Stereochemistry</i>	CO5	4
7	<i>Organic reaction and synthesis of drug molecules</i>	CO6	4

LESSON PLAN

S N	Topics to be covered	Teaching Aids	Reference
<i>Unit-1: Atomic and molecular Structure</i>			
L1	Schrodinger equation	Chalkboard	T2
L2	Particle in 1D box solution and their applications	Chalkboard	T2
L3	Molecular orbital theory for some diatomic molecules.	ppt	T2,T3
L4	Pi-molecular orbital diagram of butadiene and benzene, aromaticity	Chalkboard & ppt	T2,T3
L5	Crystal field theory for transition metal ions.	Chalkboard & ppt	T1,T2
L6	Energy level diagram of transitional metal ions & magnetic properties	Chalkboard	T1
L7	Band structure of solid & role of doping.	Chalkboard	T9,T10
<i>Unit-2: Spectroscopic techniques and applications</i>			
L8	Principle of spectroscopy & selection rule	Chalkboard& ppt	T4, T3
L9	Electronic &UV spectroscopy.	Chalkboard& ppt	T4, T3
L10	Vibrational and rotational spectroscopy.	Chalkboard& ppt	T4, T3
11	Fluorescence and its applications in medicine,	Chalkboard& ppt	T4, T3
12	NMR spectroscopy	Chalkboard& ppt	T4, T3
13	Surface characterizations techniques	Chalkboard& ppt	T4, T3
<i>Unit-3: Intermolecular forces</i>			
14	Ionic, dipolar and van Der Waals interactions	Chalkboard	T2, T3
15	Equation of state of real gases and critical	Chalkboard	T2, T3

	phenomenon		
<i>Unit-4: Use of free energy in chemical equilibrium</i>			
16	First and second law of thermodynamics & their functions.	Chalkboard	T9, T10
17	Estimation of entropy and free energies	Chalkboard	T9, T10
18	Cell potential, Nernst equation and applications	Chalkboard	T9, T10
19	Acid-base, oxidation-reduction & solubility equilibrium	Chalkboard	T6, T7
20	Water chemistry,	Chalkboard	T6, T7
21	corrosion, Ellingham diagrams		T6, T7
<i>Unit-5: Periodic properties</i>			
22	Effective nuclear charge, variations of s, p, d, f orbital energy of atoms in periodic table.	Ppt presentation	T1, T3
23	Electronic configuration, ionic size, ionization energy, electron affinity, ionization energy, polarizability	Ppt presentation	T1, T3
24	Co-ordination compound, HSAB concept	Ppt presentation	T1, T3
25	VSEPR theory and molecular geometry	Ppt presentation	T1, T3
<i>Unit-6: Stereochemistry</i>			
26	Representation of 3D structures, structural isomer & stereoisomer	Chalkboard	T3, T5
27	Configurations, symmetry and chirality of molecule	Chalkboard	T3, T5
28	optical activity, enantiomers, diastereomers	Chalkboard	T3, T5
29	Absolute configurations, conformational analysis and isomerism in transitional metal compounds	Chalkboard	T3, T5
<i>Unit-7: Organic reaction and synthesis of drug molecules</i>			
30	Reaction mechanism: substitution, addition reactions	Chalkboard	T8, T9, T10
31	Elimination (E1 and E2) reactions	Chalkboard	T8, T9, T10
32	Oxidations and reductions, cyclization and ring opening	Chalkboard	T3, T8
33	Synthesis of some commonly used drug molecules	Chalkboard	T3

Reference books:

T1: Concise Inorganic Chemistry, 4th edition, J.D. Lee, Chapman & Hall

T2: Physical Chemistry, P. C. Rakshit, Sarat Book House.

T3: Engineering Chemistry, Rath and Chakraborty

T4: Fundamental of molecular spectroscopy, by C. N. Banwell

T5: Stereochemistry of Organic Compounds, by D. Nasipuri.

T6: Jain & Jain, engineering Chemistry, 15th edition, Dhanpat Rai Publishing Company, 2008.

T7: Shashi Chawla, A Text book of Engg Chemistry, Dhanpat Rai & Company, 2003.

T8: Peter Sykes, A Guidebook Mechanism in Organic Chemistry, 6th ed. Orient Longman.

T9: S. Bandyopadhyay & N.K. Hazra, Fundamentals engg chemistry, Chhaya Prakashani Pvt. Ltd.

T10: Gourkrishna Dasmohapatra, Engineering Chemistry, 2nd edition, Vikas Publication.

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Signature of Faculty